Let's learn more QM from the simplest problem

- (b) Looking into the Physics of the solutions
 - $E_n = \frac{n^2 \pi^2 h^2}{2ma^2}$ discrete energies (selected by physics)

 $\propto \frac{1}{m}$ and $\propto \frac{1}{n^2}$

"Confinement" leads to discrete energies

size a in which particle lives

[behind the scene it is the boundary conditions due to confinement]

that matter

Quantum Size Effects: Beginning of Nanoscience and Technology

E.g. absorption/emission of nanoparticles tunable by size (why?)

⇒ smaller mass particle has more noticeable discrete - energy effect Where are discrete energies noticeable?

Atoms! Atomic spectrum

lightest - electrons confined to vicinity of nucleus

 $m \rightarrow me$ (electron mass) $\frac{\hbar^2}{me} \approx 7.62 \text{ eV} \cdot \text{Å}^2$ Keep number in mind

Size ~ Ă

°° Particle-in-a-ID-box predicts energies (energy separation) of order of $\frac{\hbar^2}{m_{e,D^2}} \approx a$ few eV (which is right!)

- * Atomic energies are always about eV to tens of eV
- " ID Well is a simple picture to think about atoms
- " Could consider 2D/3D boxes/wells
 - · more realistic OK
 - " but $\frac{\hbar^2}{ma^2}$ is always there in the results
 - > the same energy order!

• Engineering absorption

energy

energy

absorption $E_{2}-E_{1}=hV_{lower}$ energy $E_{2}-E_{1}'=hV_{higher}$

Someone comes to you and ask: I want absorption at 7.13 eV.

Can you design a system delivering this specification?

Yes! This is done using semiconductors.

· Extend application to neutron/proton confined in sucleus. (Ex.)

(c) Picking up better sense on Quantum Mechanics

(i) Why does
$$\forall i = \sqrt{\frac{2}{a}} \sin(\frac{\pi x}{a})$$
 have lower energy than $\forall i = \sqrt{\frac{2}{a}} \sin(\frac{\pi \pi x}{a})$?

Inside the box: $-\frac{h^2}{2m} \frac{d^2}{dx^2} \psi = E \psi$ 7 ISE

(U=0 in box)

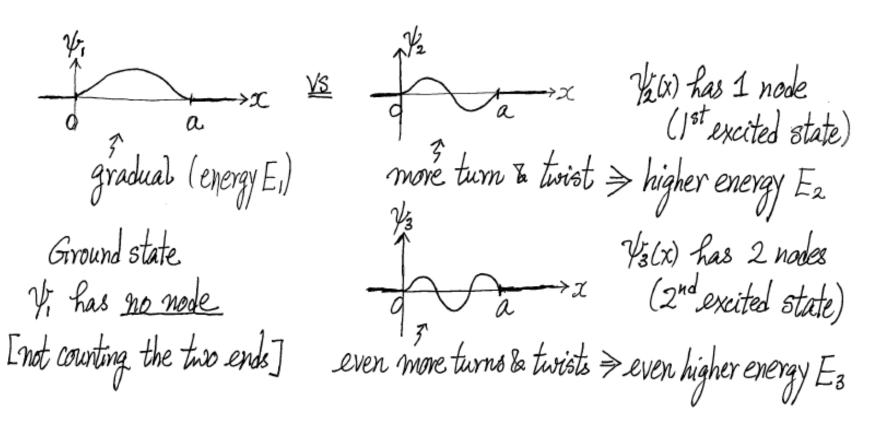
kinetic energy term $\sim \frac{d^2}{dx^2} \psi$

Same range "a": more conjegling around

(more turns/twist)

 \Rightarrow higher kinetic energy

This is true even when $U(x) \neq 0$



How about $V_{48}(x)$? 47 nodes (47th excited state) E48? $V_{48}(x)$? Sketch it Q: Could we have guessed what 4/260) is like? How about How about

this is as gradual as V can be and be Continuous!

turn around in a maller interval bigger $\frac{\int^2 \psi}{\int x^2}$ higher energy No good! More gradual: Looks good. But discontinuous!

(ii) "Think like a Computer"

Another way to understand why Boundary Conditions select specific energies In solving TISE, at one point we have (after considering 1/10) = 0)

$$\gamma(x) = A \sin kx$$
 (0E = \frac{\hbar^2 k^2}{2m}

Why couldn't we have $E < E_1 = \frac{\pi^2 h^2}{2ma^2}$ (Ground state energy)?

energy

Let's try different values of E and see what happens

Pick $E \Rightarrow try \ k = \int \frac{2mE}{t^2}$

Noesn't connect at x=a Try a E≥0 (< E1), k≈0 meaning Not QM acceptable Try a higher one (E < E,), Until $E = E_1$, first acceptable energy connects to \(\psi(a)=0 giving physically well-behaved wavefunction

Try
$$E_1 < E < E_2 \left(= \frac{4 \pi^2 h^2}{2ma^2} \right)$$

doesn't connect (out!)

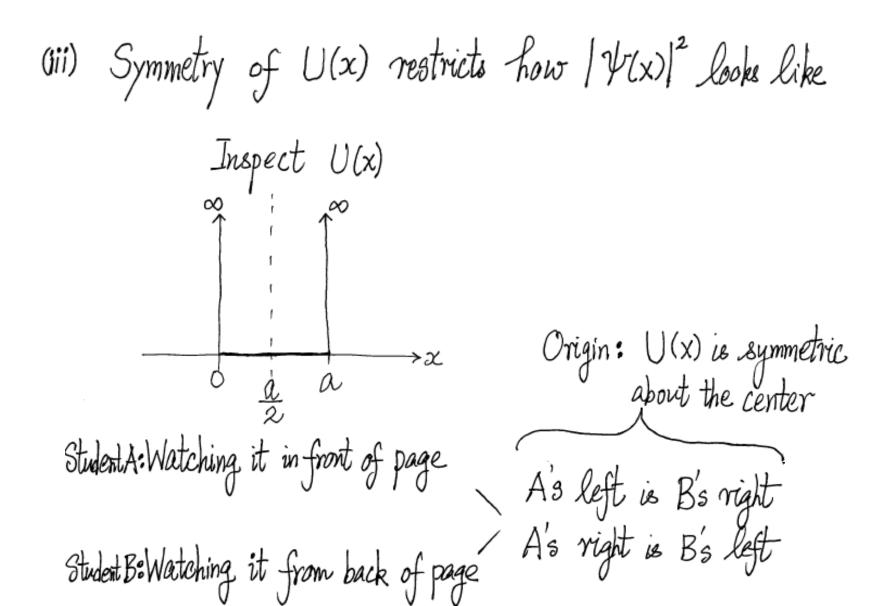
Until E = E2, second acceptable energy
giving physically
well-behaved wavefunction

and so on ...

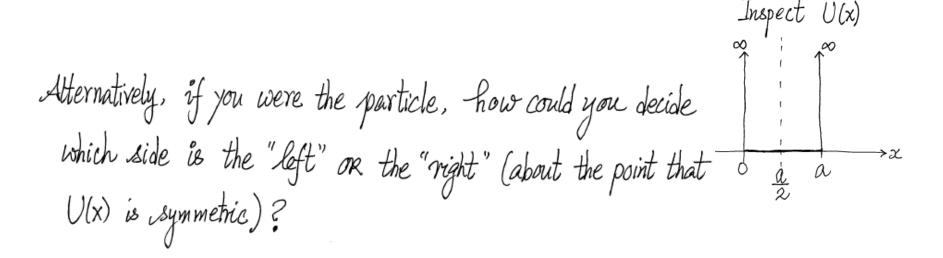
· Think like a computer hope understand how B.C. works to select E

The idea here is general, although we are discuss 1D box problem

Better strategy: Let computer think like a computer!
This is how a program solving TISE numerically works.



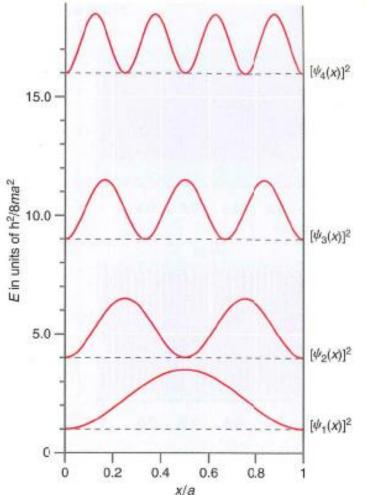
If A says the probability of finding the particle is more (biased) on the left side (if it were true), B will say it is more on the right side. There are <u>contradictory views</u>.



 $|\Psi(x)|^2$ should be <u>symmetric</u> about the point that U(x) is symmetric physical interpretation of Ψ is about $|\Psi|^2$ (not Ψ itself)

Inspect $|Y_n(x)|^2$

It is symmetric about the midpoint of the box/well (or x/a = 0.5)

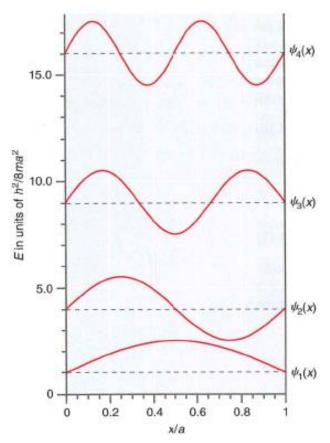


Implication on Y(x)?

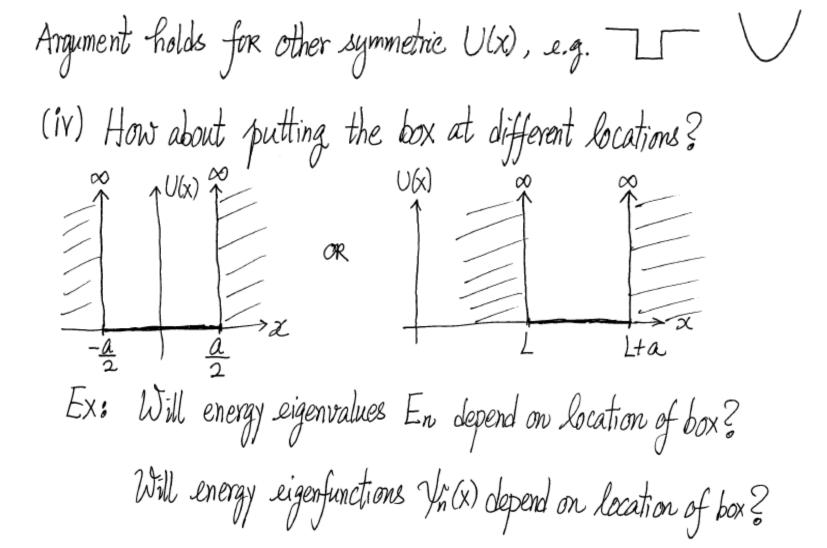
Vn(x) can either be symmetric or anti-symmetric

tor a box in 0<x<a, $\psi(x) = \psi(a-x)$ symmetric (for $0 < x < \frac{a}{2}$) $\psi(x) = -\psi(a-x)$ antisymmetric (for $0 < x < \frac{a}{2}$) Could have also guessed the form of other Vn's.

Inspect that the anti-symmetric (or odd) ones must be zero at the center of the box (point about which U(x) is symmetric). Why?



Important to note that the argument here is general for symmetric U(x), not only for the 1D box/well



(v) Bound States and bound states can be normalized What are bound states?

Generally. $V(x) \rightarrow 0$ as $x \rightarrow \pm \infty$ [so all energy eigenfunctions of 1D box are bound states] (but NOT all energy eigenfunctions of other QM problems are bound states)

If $\psi_E(x)$ is an energy eigenfunction, a state of definite energy E, then it is a bound state if $E < U(x \to \pm \infty)$.

Bound states can always be normalized

Another implication:

There are energy eigenfunctions that are not bound states. They cannot be normalized according to $\int |V(x)|^2 dx = 1$ all space

Their "normalization" is handled differently

E.g. free particle in $-\infty < \infty < +\infty$

The wavefunctions are plane wave of the form exp(ikx)